

Tetrahedron 58 (2002) 9983-9988

TETRAHEDRON

Manganese(III) acetate based tandem oxidation of various α and β -alkoxy α , β -unsaturated ketones

Cihangir Tanyeli,* Devrim Özdemirhan and Bengü Sezen

Department of Chemistry, Middle East Technical University, 06531 Ankara, Turkey

Received 25 June 2002; revised 30 September 2002; accepted 17 October 2002

Abstract—We described the unusual results of manganese(III) acetate based tandem oxidation of various α and β -alkoxy α , β -unsaturated ketones to afford the corresponding α' -acetoxy- α' -phenyl substituted oxidation products in good yields. The tandem oxidation to monoacetoxylation ratio can be tuned by the amount of manganese(III) acetate. © 2002 Elsevier Science Ltd. All rights reserved.

1. Introduction

In the last decade, Manganese(III)-based oxidative freeradical reactions have been developed into a versatile protocol for the formation of highly functionalised products from simple precursors.^{1–10} In 1976, Williams and Hunter reported that Mn(OAc)₃·2H₂O oxidation of enones in AcOH at reflux affords α' -acetoxy enones in low yields.¹¹ In connection with our synthetic studies with manganese(III) acetate,¹² we found that trapping the α' -keto radicals obtained from β -alkoxy α , β -unsaturated ketones by benzene used as a solvent is much faster than acetoxylation, affording good yields of tandem oxidation products.^{12d} These interesting results prompted us towards the development of a new method in the field of the regioselective tandem oxidation of α -alkoxy and β -alkoxy α , β -unsaturated ketones. Here, we report our results obtained from the Mn(OAc)₃ oxidation of various α -alkoxy and β -alkoxy α , β unsaturated ketones and an optimization of the reaction conditions.

2. Results and discussion

As a starting point, various β -alkoxy α , β -unsaturated ketones **1** were prepared from the corresponding 1,3diketones using slightly modified literature procedures^{13,14} and were then allowed to react with Mn(OAc)₃. We have found that Mn(III)-based oxidation of β -alkoxy α , β unsaturated ketones is infact a versatile synthetic procedure with broad applicability.¹⁵ For instance, reaction of **1a** with 3 equiv. of Mn(OAc)₃ in benzene as the solvent for 8 h at reflux afforded 61% of tandem oxidation product **2a**.



Scheme 1.

Characterization of the product revealed the introduction of a phenyl group together with an acetoxy moiety on the α' position (Scheme 1). The full results are summarized in Table 1.

The reaction presumably proceeds via the formation of the Mn(III) enolate 3, which loses Mn(II) upon one-electron oxidation to give α' -keto radical $4^{6,10d,16,17}$ (Scheme 2). Addition of this radical to benzene yields another radical, 5. One-electron oxidation of the resultant radical 5 gives α' phenyl β -alkoxy α , β -unsaturated ketones 6. The introduction of aryl groups to ketones was reported to be confined to acetone, by Vinogradov et al.,¹⁸ since α' -keto radicals generated by higher ketones would result in secondary radicals which would dimerize, or tertiary radicals that are prone to further oxidation. In contrast to this conclusion, in this study, no dimerization was observed. Oxidation of intermediate 6 by another equivalent of Mn(OAc)₃ provides α' -acetoxy α' -phenyl β -alkoxy α,β -unsaturated ketones 2. Similarly, cyclopentenones 1b and 1c (entries 2 and 3) give 73 and 68% of tandem oxidation products 2b and 2c, respectively, together with a small amount of recovered starting material and some unidentified polymeric products.

With cyclohexenone derivatives 1d-h, acetoxylation at the α' -position competes effectively with the tandem oxidation. For instance, reaction of 3-ethoxy-2-cyclohexenone 1d with 3 equiv. of Mn(OAc)₃ in benzene for 8 h at reflux gives 2d in 56% yield, 7d in 35% yield and monophenylated product

Keywords: enones; manganese and compounds; oxidation.

^{*} Corresponding author. Tel.: +90-312-210-32-22; fax: +90-312-210-12-80; e-mail: tanyeli@metu.edu.tr

Entry	Reactant	Product	2	Yield (%)	Product	7	Yield (%)	Time (h)
1	EtO	O OAc Ph EtO	a	61		_		8
2	EtO	O OAc Ph EtO	b	73		-		8
3	EtO	O OAc Ph EtO	с	68		-		8
4	Eto	eto OAc Ph	d	56	Eto OAc	d	35	12
5	EtO	eto OAc	e	59	OAc Eto	e	28	12
6 ^a	Eto	O OAc Ph	f	62	Eto OAc	f	26	11
7 ^b	Eto Ph	Eto Ph	g	65	Eto Ph	g	25	10
8	Eto	eto OAc	h	60	Eto OAc	h	31	12
9 ^c	OEt O	OEt O OAc Ph	i	58	OEt O OAc	i	35	12

Table 1.	Tandem	oxidation of	B-alkoxy	α .B-unsaturated	ketones v	vith Mn(OAc) ₂	in benzene
Table 1.	randem	oxidation of	pankony	, u,p unsaturated	Rectones v	vitil iville	Oners.	in benzene

^a The reaction was carried out with racemic mixture of 1f. 2f was isolated as 6:4 diastereomeric mixture, while 7f was isolated as 8:2 E/Z mixture and 6f as only E isomer by using ¹H NMR data. ^b **2g** was isolated as 7:3 diastereomeric mixture. Only isolable product **7g** has *E* configuration which was determined from ¹H NMR. ^c **2i** was isolated as 7:3 diastereomeric mixture.



F (D ()	D 1 (0	V: 11(0)	D 1 (10	V: 11(0()	T ' (1)
Entry	Reactant	Product	9	Yield (%)	Product	10	Y 1eld (%)	Time (h)
1	EtO	Eto Ph	a	72	-			8
2	EtO	EIO OAc Ph	b	75	_			8
3	EtO.	EtO OAc Ph	с	63	EtOOAc	с	28	12
4	EtO	Eto OAc	d	59	Eto OAc	d	32	10

Table 2. Tandem oxidation of α -alkoxy α , β -unsaturated ketones with Mn(OAc)₃ in benzene

6d in 5% yield. The isolation of monophenylated product was also achieved in entry 6 as **6f** with 6% yield. As a model study, monoacetoxylation product **7d** was subjected to $Mn(OAc)_3$ oxidation in a similar manner in entry 4 (3 equiv. $Mn(OAc)_3$ in benzene). As expected from the proposed reaction mechanism, no product formation was observed. When 4 equiv. of $Mn(OAc)_3$ was used (entry 4), we observed drastic increase in the chemical yield of monoacetoxylated product **7d** (53%) and a decrease in the tandem oxidation product **2d** (31%). Similar results were obtained for entries 5–8. Hence, we decided that the critical amount of oxidant for six-membered enones was 3 equiv.

Notably Z-3-ethoxy-2-pentenone (entry 9) behaves very similarly, giving 2i in 58% yield as a 7:3 diastereomeric mixture. 35% of monoacetoxylated product 7i was also isolated.

In the next step of this study, 3-ethoxy-2-cyclopentenone (entry 1) was chosen as a model compound. The Mn(OAc)₃ oxidation was performed in various solvents. With nitrobenzene and pyridine, no product formation was observed. With anisole, aniline and furan, only uncharacterized oxidation products were obtained. Bromobenzene and chlorobenzene each yielded 5-acetoxy-3-ethoxy-5-phenyl-2-cyclopentenone in 38 and 12% yields, respectively.

Other substrates various α -alkoxy α , β -unsaturated ketones **8** were next prepared from the corresponding 1,2-diketones using slightly modified literature procedures.¹⁵ On reaction of **8a** with 3 equiv. of Mn(OAc)₃ in benzene at reflux for 8 h, **9a** was produced in 72% yield. Further examples of the reaction are summarized in Table 2. We suggest that reactions follow a similar route to that obtained in Scheme 2.

In conclusion, we have discovered an interesting and potentially useful $Mn(OAc)_3$ based tandem oxidation reaction. The applicability of a one-pot double oxidation method is attractive for further use in synthesis.

3. Experimental

¹H and ¹³C NMR spectra were recorded in CDCl₃ on Bruker Spectrospin Avance DPX 400 spectrometers. Chemical shifts are given in ppm from tetramethylsilane. Coupling constants (*J*) were recorded in Hertz. IR spectra were obtained from a Perkin–Elmer Model 1600 series FT–IR spectrometer and are reported in cm⁻¹. Mass spectra were recorded on a Varian MAT 212. Thin layer chromatography (TLC) was performed on Merck plastic-backed silica gel plates. Compounds were visualised under a UV lamp. Column chromatography was performed using silica gel (Flash Silica 60, 32–63 µm). 1,2- and 1,3-Diketones were purchased from Aldrich. α-Ethoxy and β-ethoxy α,βunsaturated compounds were synthesized according to literature procedure.^{13,14}

3.1. General procedure for the $Mn(OAc)_3$ oxidation of β -alkoxy α,β -unsaturated ketones

A mixture of $Mn(OAc)_3$ (2.41 g, 9.0 mmol) in benzene (or corresponding solvent) (100 mL) was refluxed for 45 min. under a Dean–Stark trap, then cooled to room temperature. β -Alkoxy α , β -unsaturated ketone (3.0 mmol) was gradually added and the mixture was allowed to reflux until the dark brown colour disappeared (TLC monitoring). The reaction mixture was diluted with ethyl acetate (100 mL) and the organic phase was washed with 1N HCl (100 mL), saturated NaHCO₃ and brine. The organic phase was dried over MgSO₄ and evaporated in vacuo. The crude product was separated by flash column chromatography using ethyl acetate/hexane as eluent to afford the product.

3.1.1. 5-Acetoxy-3-ethoxy-5-phenyl-2-cyclopentenone 2a. (0.48 g, 61%) as a white solid, mp 94–96°C; $R_{\rm f}$ (EtOAc/hexane 1:3) 0.35; $\nu_{\rm max}$ (neat) 1740, 1715 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.18–7.34 (5H, m, Ph), 5.25 (1H, s, CH=), 4.02–4.13 (2H, m, MeCH₂O), 3.30 (1H, d, J=18 Hz, CH_aH_bCOAcPh), 3.13 (1H, d, J=18 Hz, CH_aH_bCOAcPh), 3.13 (1H, d, J=18 Hz, CH_aH_bCOAcPh), 2.10 (3H, s, *Me*CO₂), 1.36 (3H, t, J=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 199.4, 185.8, 170.1, 138.6, 129.0, 128.5, 125.1, 102.5, 84.1, 68.2, 43.0, 21.7, 14.5; m/z (EI) 260 (9), 218 (32), 217 (28), 200 (25), 105 (100), 77 (22%). HRMS (EI): M⁺, found 260.1049. C₁₅H₁₆O₄ requires 260.1048.

3.1.2. 5-Acetoxy-3-ethoxy-2-methyl-5-phenyl-2-cyclopentenone 2b. (0.60 g, 73%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:2) 0.33; $\nu_{\rm max}$ (neat) 1743, 1714 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.13–7.40 (5H, m, Ph), 4.29 (2H, q, J=7 Hz, MeCH₂O), 3.66 (1H, d, J=17 Hz, CH_aH_b-COAcPh), 3.33 (1H, d, J=17 Hz, CH_aH_bCOAcPh), 2.19 (3H, s, *Me*CO₂), 1.71 (3H, s, *Me*C=), 1.45 (3H, t, J=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 204.9, 183.3, 170.3, 140.2, 139.1, 129.0, 127.9, 127.3, 124.9, 114.3, 84.0, 50.9, 35.5, 21.8, 15.7, 6.8; *m*/z (EI) 276 (18), 275 (83), 247 (13), 231 (11), 218 (22), 217 (100), 215 (29%). HRMS (EI): M⁺, found 274.1198. C₁₆H₁₈O₄ requires 274.1206.

3.1.3. 5-Acetoxy-3-ethoxy-2-ethyl-5-phenyl-2-cyclopentenone 2c. (0.59 g, 68%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:2) 0.35; $\nu_{\rm max}$ (neat) 1746, 1715 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.15–7.33 (5H, m, Ph), 4.24 (2H, q, J=7 Hz, MeCH₂O), 3.64 (2H, d, J=17 Hz, CH₂COAcPh), 2.22 (2H, q, J=7 Hz, MeCH₂C=), 2.18 (3H, s, $MeCO_2$), 1.43 (3H, t, J=7 Hz, $MeCH_2O$), 1.04 (3H, t, J=7 Hz, $MeCH_2C=$); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 201.1, 176.9, 170.6, 139.1, 128.3, 125.7, 124.9, 121.8, 120.5, 119.9, 84.0, 68.4, 42.2, 21.7, 15.4, 15.2, 13.1; m/z (EI) 290 (15), 289 (83), 261 (18), 232 (16), 231 (100), 229 (17%). HRMS (EI): M⁺, found 288.1371. C₁₇H₂₀O₄ requires 288.1362.

3.1.4. 6-Acetoxy-3-ethoxy-6-phenyl-2-cyclohexenone 2d. (0.46 g, 56%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:8) 0.81; $\nu_{\rm max}$ (neat) 3066, 1758, 1675 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.19–7.38 (5H, m, Ph), 5.56 (1H, s, CH=), 3.83–3.95 (2H, m, MeCH₂O), 2.10–2.30 (4H, m, CH₂CH₂-COAcPh), 2.05 (3H, s, *Me*CO₂), 1.29 (3H, t, *J*=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 199.3, 176.6, 169.9, 137.6, 128.6, 128.5, 128.3, 126.8, 126.7, 103.2, 96.1, 64.9, 29.4, 28.2, 21.8, 14.1; HRMS (EI): M⁺, found 274.1211. C₁₆H₁₈O₄ requires 274.1206.

3.1.5. 6-Acetoxy-3-ethoxy-2-cyclohexenone 7d. (0.21 g, 35%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:8) 0.66; $\nu_{\rm max}$ (neat) 1736, 1665 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 5.37 (1H, s, CH=), 5.29 (1H, dd, J=5, 13 Hz, CHOAc), 3.92 (2H, q, J=7 Hz, MeCH₂O), 2.63–2.67 (1H, m, CH_aCH_bC=), 2.52–2.54 (1H, m, CH_aCH_bC=), 2.20–2.24 (1H, m, CH_aCH_bCOAc), 2.19 (3H, s, *Me*OAc), 2.07–2.16 (1H, m, CH_aCH_bCOAc), 1.37 (3H, t, J=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 193.4, 177.1, 170.5, 101.5, 72.8, 65.2, 28.3 27.1, 21.2, 14.4; *m*/*z* (EI) 217 (19), 200 (14), 199 (100%); HRMS (EI): M⁺, found 198.0883. C₁₀H₁₄O₄ requires 198.0892.

3.1.6. 3-Ethoxy-6-phenyl-2-cyclohexenone 6d. (0.03 g, 5%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:8) 0.72; $\nu_{\rm max}$ (neat) 3150, 1691, 1641 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.10–7.28 (5H, m, Ph), 5.44 (1H, s, CH=), 3.88 (2H, q, J=7 Hz, MeCH₂O), 3.44–3.49 (1H, dd, J=5, 10 Hz, CHPh), 2.40–2.51 (2H, m, =CCH₂), 2.12–2.17 (2H, m, CH₂CH₂CHPh), 1.32 (3H, t, J=7 Hz, MeCH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 199.6, 177.7, 140.3, 128.9, 128.7, 127.6, 127.2, 126.6,

103.6, 64.8, 52.3, 29.8, 28.6, 14.5; HRMS (EI): M^+ , found 216.1147. $C_{14}H_{16}O_2$ requires 216.1151.

3.1.7. 6-Acetoxy-3-ethoxy-2-methyl-6-phenyl-2-cyclohexenone 2e. (0.51 g, 59%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:2) 0.66; $\nu_{\rm max}$ (neat) 3020, 2975, 1725, 1695 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.14–7.33 (5H, m, Ph), 4.01 (2H, q, J=7 Hz, MeCH₂O), 2.16–2.61 (4H, m, CH₂CH₂), 2.14 (3H, s, MeOAc), 1.78 (3H, s, MeC=), 1.37 (3H, t, J=7 Hz, MeCH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 195.0, 177.2, 170.5, 137.2, 129.3, 128.5, 127.9, 127.1, 126.9, 121.1, 87.4, 65.5, 59.2, 47.3, 22.1, 20.8, 14.5; HRMS (EI): M⁺, found 288.1359. C₁₇H₂₀O₄ requires 288.1362.

3.1.8. 6-Acetoxy-3-ethoxy-2-methyl-2-cyclohexenone 7e. (0.18 g, 28%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:2) 0.51; $\nu_{\rm max}$ (neat) 1733, 1658 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 5.90–5.91 (1H, m, CHOAc), 4.04 (2H, q, *J*=7 Hz, MeCH₂O), 2.54–2.63 (1H, m, =CCH_aH_bCH₂), 2.34– 2.39 (1H, m, =CH_aH_bCH₂), 2.15–2.17 (2H, m, CH₂CH₂-COAc), 2.14 (3H, s, *Me*CO₂), 1.76 (3H, s, *Me*C=), 1.33 (3H, t, *J*=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 198.7, 170.6, 164.8, 120.1, 70.1, 64.7, 32.1, 27.0, 21.2, 15.9, 15.7; *m*/z (EI) 213 (100), 153 (11%). HRMS (EI): M⁺, found 212.1052. C₁₁H₁₆O₄ requires 212.1049.

3.1.9. 6-Acetoxy-3-ethoxy-5-methyl-6-phenyl-2-cyclohexenone 2f. (0.54 g, 62%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:4) 0.24; $\nu_{\rm max}$ (neat) 3016, 2976, 1728, 1685 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) (6:4 diastereomeric mixture) 7.01–7.30 (2×5H, m, Ph), 5.43 (1H, s, CH=), 5.36 (1H, s, CH=), 3.82–3.96 (2×2H, m, MeCH₂O), 2.43–2.48 (2×1H, m, CHMe), 2.23–2.36 (2×2H, m, CH₂CHMe), 2.11 (3H, s, *Me*CO₂), 2.10 (3H, s, *Me*CO₂), 1.36 (3H, t, *J*=7 Hz, *Me*CH₂O), 1.35 (3H, t, *J*=7 Hz, *Me*CH₂O), 0.91 (3H, d, *J*=6 Hz, *Me*CH), 0.84 (3H, d, *J*=6 Hz, *Me*CH); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 195.3, 193.5, 178.2, 177.4, 170.2, 166.5, 140.1, 138.9, 129.5, 128.8, 128.5, 128.3, 127.2, 103.2, 101.3, 64.7, 63.5, 61.4, 60.8, 38.1, 37.5, 35.7, 34.8, 31.3, 30.8, 21.1, 20.6, 14.7, 14.6; HRMS (EI): M⁺, found 288.1356. C₁₇H₂₀O₄ requires 288.1362.

3.1.10. 6-Acetoxy-3-ethoxy-5-methyl-2-cyclohexenone 7f. (0.17 g, 26%) as a colourless oil; R_f (EtOAc/hexane 1:4) 0.21; ν_{max} (neat) 1739, 1646 cm⁻¹; δ_H (400 MHz, CDCl₃) (8:2 *E*/Z mixture) 5.46 (1H, d, *J*=5 Hz, CH=), 5.40 (1H, s, CH=), 5.38 (1H, s, CHOAc), 5.09 (1H, d, *J*=12 Hz, CHOAc), 3.89-4.02 (2×2H, m, MeCH₂O), 2.83-2.89 (2×1H, m, CHMe), 2.45-2.61 (2×1H, m, =CCH_aH_b), 2.34-2.43 (2×1H, m, =CCH_aCH_b), 2.24 (3H, s, *Me*CO₂), 2.23 (3H, s, MeCO₂), 1.39-1.43 (2×3H, m, *Me*CH₂O), 1.12 (3H, d, *J*=7 Hz, *Me*CH), 1.09 (3H, d, *J*=7 Hz, *Me*CH); δ_C (100.6 MHz, CDCl₃) 193.2, 192.7, 176.1, 174.9, 170.9, 170.5, 102.7, 101.4, 101.0, 78.3, 75.8, 65.2, 64.6, 45.5, 37.6, 36.9, 35.7, 33.8, 31.9, 21.2, 21.1, 18.1, 14.4, 13.7; HRMS (EI): M⁺, found 212.1042. C₁₁H₁₆O₄ requires 212.1049.

3.1.11. 3-Ethoxy-5-methyl-6-phenyl-2-cyclohexenone 6f. (0.04 g, 6%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:4) 0.26; $\nu_{\rm max}$ (neat) 3210, 1693, 1643 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.03–7.28 (5H, m, Ph), 5.45 (1H, s, CH=), 3.80–3.92 (2H, m, MeCH₂O), 3.06 (1H, d, *J*=11 Hz, CHPh), 2.44–2.49 (1H, dd, *J*=4, 16 Hz, CH_aH_bCHMe), 2.32–2.41

(2H, m, CH_aH_bCHMe), 1.33 (3H, t, J=7 Hz, MeCH₂O), 0.84 (3H, d, J=6 Hz, MeCH); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 198.7, 176.7, 139.7, 128.6, 128.3, 127.9, 127.4, 126.3, 101.3, 65.4, 53.2, 30.2, 21.3, 15.2, 14.3; HRMS (EI): M⁺, found 230.1302. C₁₅H₁₈O₂ requires 230.1307.

3.1.12. 6-Acetoxy-3-ethoxy-5,6-diphenyl-2-cyclohexenone **2g.** (0.68 g, 65%) as a colourless oil; $R_{\rm f}$ (EtOAc/ hexane 1:3) 0.42; $\nu_{\rm max}$ (neat) 3010, 2973, 1738, 1674 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) (7:3 diastereomeric mixture) 6.93– 7.16 (10H, m, Ph), 5.71 (1H, s, CH=), 5.60 (1H, s, CH=), 3.75–3.83 (2×3H, m, MeCH₂O), 2.33–2.40 (2×2H, m, CH₂CHPh), 1.91 (3H, s, *Me*CO₂), 1.89 (3H, s, *Me*CO₂), 1.13 (3H, t, *J*=7 Hz, *Me*CH₂O), 1.09 (3H, t, *J*=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 196.7, 194.2, 176.1, 175.7, 172.3, 171.6, 140.4, 139.8, 129.1, 127.9, 127.7, 126.5, 109.8, 102.0, 101.3, 65.3, 64.8, 42.4, 41.2, 36.2, 35.4, 20.1, 19.8, 14.3, 14.1; HRMS (EI): M⁺, found 350.1531. C₂₂H₂₂O₄ requires 350.1518.

3.1.13. 6-Acetoxy-3-ethoxy-5-phenyl-2-cyclohexenone 7g. (0.21 g, 25%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:3) 0.37; $\nu_{\rm max}$ (neat) 3015, 2875, 1734 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.24–7.36 (5H, m, Ph), 5.63 (1H, d, J=13 Hz, CHOAc), 5.46 (1H, s, CH=), 3.96 (2H, q, J=7 Hz, MeCH₂O), 3.50 (1H, dt, J=5, 13 Hz, CHPh), 2.87 (1H, dd, J=6, 18 Hz, CH_aH_bCHPh), 2.70 (1H, dd, J=5, 18 Hz, CH_aH_bCHPh), 1.95 (3H, s, MeCO₂), 1.33 (3H, t, J=7 Hz, MeCH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 192.8, 175.8, 170.6, 139.6, 129.1, 128.8, 128.5, 127.9, 127.7, 101.4, 76.0, 65.5, 45.0, 37.3, 20.9, 14.4; m/z (EI) 274 (100), 215 (22%); HRMS (EI): M⁺, found 274.1193. C₁₆H₁₈O₄ requires 274.1206.

3.1.14. 6-Acetoxy-3-ethoxy-4,4-dimethyl-6-phenyl-2cyclohexenone 2h. (0.54 g, 60%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:5) 0.68; $\nu_{\rm max}$ (neat) 3012, 2970, 1742, 1682 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.09–7.28 (5H, m, Ph), 5.50 (1H, s, CH=), 3.82–4.05 (2H, m, MeCH₂O), 2.17 (1H, d, J=13 Hz, CH_aH_b COAcPh), 2.10 (3H, s, $MeCO_2$), 1.87 (1H, d, J=13 Hz, CH_aH_b COAcPh), 1.32 (3H, t, J=7 Hz, MeCH₂O), 1.29 (3H, s, MeMeC), 1.14 (3H, s, MeMeC); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 192.8, 175.7, 170.6, 139.6, 129.1, 128.5, 127.9, 127.7, 126.8, 101.4, 96.5, 76.0, 65.5, 45.0, 37.3, 27.7, 20.9, 14.5; HRMS (EI): M⁺, found 302.1511. C₁₈H₂₂O₄ requires 302.1518.

3.1.15. 6-Acetoxy-3-ethoxy-4,4-dimethyl-2-cyclohexenone 7h. (0.21 g, 31%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:5) 0.60; $\nu_{\rm max}$ (neat) 1745, 1660 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 5.30 (1H, dd, *J*=7, 12 Hz, CHOAc), 5.09 (1H, s, CH=), 3.74 (2H, q, *J*=7 Hz, MeCH₂O), 2.01 (3H, s, *Me*CO₂), 1.84–1.90 (2H, m, CH₂CHOAc), 1.21 (3H, t, *J*=7 Hz, *Me*CH₂O), 1.19 (3H, s, *Me*MeC), 1.06 (3H, s, MeMeC); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 193.7, 182.2, 170.7, 99.8, 70.9, 65.2, 41.7, 37.7, 27.7, 25.8, 21.3, 14.3; *m/z* (EI) 227 (100), 215 (7%); HRMS (EI): M⁺, found 226.1212, C₁₂H₁₈O₄ requires 226.1205.

3.1.16. 1-Acetoxy-4-ethoxy-1-phenyl-pent-3-en-2-one 2i. (0.46 g, 58%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:10) 0.90; $\nu_{\rm max}$ (neat) 3015, 2969, 1742, 1692 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) (7:3 diastereoisomeric mixture) 6.84–

7.56 (2×5H, m, Ph), 6.39 (2×1H, s, CHOAcPh), 6.33 (1H, s, CH=), 6.32 (1H, s, CH=), 3.87–3.93 (2×2H, m, MeCH₂O), 2.19 (3H, s, *Me*CO₂), 2.15 (3H, s, *Me*CO₂), 2.05 (3H, s, *Me*C=), 1.89 (3H, s, *Me*C=), 1.29–1.37 (2×3H, m, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 205.2, 201.0, 169.9, 168.6, 159.8, 159.2, 137.4, 137.1, 135.2, 133.3, 130.8, 130.0, 129.0, 128.9, 128.7, 127.3, 127.1, 114.1, 113.9, 90.1, 81.9, 63.4, 63.2, 61.0, 21.2, 20.8, 20.0, 15.3, 15.2; HRMS (EI): M⁺, found 262.1198, C₁₅H₁₈O₄ requires 262.1205.

3.1.17. 1-Acetoxy-4-ethoxy-pent-3-en-2-one 7i. (0.20 g, 35%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:10) 0.83; $\nu_{\rm max}$ (neat) 1735, 1685 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 6.44 (1H, s, CH=), 4.78 (2H, s, CH₂OAc), 3.92 (2H, q, J=7 Hz, MeCH₂O), 2.18 (3H, s, MeCO₂), 2.09 (3H, s, MeC=), 1.33 (3H, t, J=7 Hz, MeCH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 202.4, 170.2, 159.9, 136.3, 130.8, 114.1, 69.6, 63.5, 20.8, 19.8, 15.2; HRMS (EI): M⁺, found 186.2051. C₉H₁₄O₄ requires 186.0892.

3.2. General procedure for the $Mn(OAc)_3$ oxidation of $\alpha\text{-alkoxy }\alpha,\beta\text{-unsaturated ketones}$

A mixture of $Mn(OAc)_3$ (2.41 g, 9.0 mmol) in benzene (or corresponding solvent) (100 mL) was refluxed for 45 min. under a Dean–Stark trap, then cooled to room temperature. α -Alkoxy α , β -unsaturated ketone (3.0 mmol) was gradually added and the mixture was allowed to reflux until the dark brown colour disappeared (TLC monitoring). The reaction mixture was diluted with ethyl acetate (100 mL) and the organic phase was washed with 1N HCl (100 mL), saturated NaHCO₃ and brine. The organic phase was dried over MgSO₄ and evaporated in vacuo. The crude product was separated by flash column chromatography using ethyl acetate/hexane as eluent to afford the product.

3.2.1. 5-Acetoxy-2-ethoxy-3-methyl-5-phenyl-2-cylopentenone 9a. (0.59 g, 72%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:6) 0.63; $\nu_{\rm max}$ (neat) 1748, 1640 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.06–7.46 (5H, m, Ph), 4.15 (2H, q, J=7 Hz, MeCH₂O), 2.91 (1H, d, J=18 Hz, CH_aH_b-COAcPh), 2.46 (1H, d, J=18 Hz, CH_aH_bCOAcPh), 2.06 (3H, s, MeCO₂), 2.00 (3H, s, MeC=), 1.21 (3H, t, J=7 Hz, MeCH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 203.1, 169.2, 154.7, 140.2, 129.2, 129.0, 127.9, 127.3, 127.1, 90.1, 66.5, 50.0, 37.6, 21.3, 16.1, 15.3; m/z (EI) 216 (60), 188 (100), 173 (27), 159 (26), 145 (28), 131 (33), 118 (37), 105 (36), 103 (35), 91 (32), 77 (33%); HRMS (EI): (M-C₂H₂O₂), found 216.1150. C₁₆H₁₈O₄ requires 216.1151.

3.2.2. 5-Acetoxy-2-ethoxy-3-ethyl-5-phenyl-2-cylopentenone 9b. (0.65 g, 75%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:5) 0.65; $\nu_{\rm max}$ (neat) 3014, 1748, 1606, 1364, 1239, 738 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 7.26–7.46 (5H, m, Ph), 4.24–4.31 (1H, m, MeCH_aH_bO), 4.08–4.15 (1H, m, MeCH_aH_bO), 3.14 (1H, d, *J*=18 Hz, CH_aH_bCOAcPh), 3.08 (1H, d, *J*=18 Hz, CH_aH_bCOAcPh), 2.53 (2H, q, *J*=8 Hz, MeCH₂C=), 2.19 (3H, s, *Me*CO₂), 1.18–1.26 (2×3H, m, *Me*CH₂O, *Me*CH₂C=); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 197.1, 170.0, 155.9, 149.0, 138.2, 129.0, 128.5, 127.1, 124.2, 122.5, 82.1, 66.5, 40.5, 22.0, 21.6, 15.9, 11.9; *m/z* (EI) 289 (6), 261 (30), 249 (60), 229 (100), 121 (30%);

HRMS (EI): M^+ , found 288.3383, $C_{17}H_{20}O_4$ requires 288.1362.

3.2.3. 6-Acetoxy-2-ethoxy-6-phenyl-2-cylohexenone 9c. (0.52 g, 63%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:8) 0.80; $\nu_{\rm max}$ (neat) 3410, 2970, 1735, 1643 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 6.50–6.98 (5H, m, Ph), 5.39–5.44 (1H, m, CH=), 4.07–4.20 (2H, m, MeCH₂O), 2.15–2.27 (2H, m, CH₂), 2.05 (3H, s, $MeCO_2$), 1.98–2.02 (1H, m, CH_aH_b), 1.81–1.87 (1H, m, CH_aH_b), 1.35 (3H, t, *J*=7 Hz, MeCH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 200.6, 170.3, 146.5, 144.2, 128.4, 122.6, 116.4, 115.6, 114.8, 112.5, 74.5, 62.7, 51.2, 28.8, 21.1, 15.6; *m/z* (EI) 275 (6), 269 (9), 241 (20), 177 (100), 167 (18), 135 (25%); HRMS (EI): M⁺, found 274.1205, C₁₆H₁₈O₄ requires 274.3117.

3.2.4. 6-Acetoxy-2-ethoxy-2-cylohexenone 10c. (0.17 g, 28%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:8) 0.66; $\nu_{\rm max}$ (neat) 3415, 2973, 1732, 1623, 1378 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 5.77–5.80 (1H, m, CH=), 5.30 (1H, dd, J=5, 13 Hz, CHOAc), 3.71 (2H, q, J=7 Hz, MeCH₂O), 2.39–2.47 (2H, m, =CCH₂), 2.13–2.18 (1H, m, CH_aH_bCHOAc), 2.10 (3H, s, *Me*CO₂), 1.99–2.08 (1H, m, CH_aH_bCHOAc), 1.29 (3H, t, J=7 Hz, *Me*CH₂O); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 189.4, 170.1, 150.5, 116.6, 64.0, 29.2, 22.1, 21.2, 15.2, 14.8; *m/z* (EI) 199 (100), 157 (5), 139 (20%); HRMS (EI): M⁺, found 198.0892. C₁₀H₁₄O₄ requires 198.2158.

3.2.5. 6-Acetoxy-2-ethoxy-3,5,5-trimethyl-6-phenyl-2cylohexenone 9d. (0.56 g, 59%) as a colourless oil; $R_{\rm f}$ (EtOAc/hexane 1:14) 0.41; $\nu_{\rm max}$ (neat) 3400, 2976, 1738, 1676 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 6.82–7.05 (5H, m, Ph), 3.69–3.74 (2H, m, MeCH₂O), 2.34 (1H, d, *J*=18 Hz, CH_aH_bCMeMe), 2.16 (1H, d, *J*=18 Hz, CH_aH_bCMeMe), 2.16 (1H, d, *J*=18 Hz, CH_aH_bCMeMe), 2.14 (3H, s, *Me*CO₂), 1.82 (3H, s, *Me*C=), 1.15 (3H, t, *J*=7 Hz, *Me*CH₂O), 0.99 (3H, s, *Me*MeC), 0.91 (3H, s, MeMeC); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 196.5, 171.3, 155.6, 142.4, 134.1, 128.8, 128.4, 127.2, 126.2, 114.7, 110.1, 61.4, 45.1, 34.3, 21.4, 20.3, 17.9, 15.7, 14.2; HRMS (EI): M⁺, found 316.1668. C₁₉H₂₄O₄ requires 316.1674.

3.2.6. 6-Acetoxy-2-ethoxy-3,5,5-trimethyl-2-cylohexenone 10d. (0.23 g, 32%) as a colourless oil; $R_{\rm f}$ (EtOAc/ hexane 1:14) 0.33; $\nu_{\rm max}$ (neat) 3417, 2988, 1730, 1632 cm⁻¹; $\delta_{\rm H}$ (400 MHz, CDCl₃) 5.08 (1H, s, CHOAc), 3.77–3.85 (1H, m, MeCH_aH_bO), 3.63–3.70 (1H, m, MeCH_aH_bO), 2.44 (1H, d, *J*=18 Hz, *CH*_aH_bCMeMe), 2.08 (3H, s, *Me*CO₂), 2.05 (1H, d, *J*=18 Hz, *CH*_aH_b-CMeMe), 1.78 (3H, s, *Me*C=), 1.14 (3H, t, *J*=7 Hz, *Me*CH₂O), 0.96 (3H, s, *Me*MeC), 0.86 (3H, s, MeMeC); $\delta_{\rm C}$ (100.6 MHz, CDCl₃) 189.6, 170.7, 146.6, 142.3, 81.2, 67.8, 44.5, 37.8, 27.7, 20.2, 19.9, 18.0, 15.8; HRMS (EI): M^+ , found 239.1281, $C_{13}H_{19}O_4$ requires 239.1284.

Acknowledgements

We thank the Middle East Technical University for the grants (no. AFP-2001-01-03-04 and AFP-2001-07-02-00-22) and the Turkish Scientific and Technical Research Council for a grant [no. TBAG-1883 (100T026)].

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